differences which although subtle are critical to an understanding of ionic conduction.

The above discussion is not intended to imply that constant-wavelength instruments are in general inferior to those based on time-of-flight. 'State-of-the-art' constant-wavelength instruments such as D2B would, we are sure, offer a considerably improved refinement to that from D1A, probably comparable in many respects to HRPD.

Finally, it is noteworthy that the X-ray structure determination (Plattner \& Völlenkle, 1979) failed to identify the displaced $\mathrm{Li}(2 \mathrm{a})$ cations or the existence of the second lithium site (4); this may be attributed to the weak scattering of X-rays by lithium ions.

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# Structural Analysis of the Threefold-Modulated Monoclinic Phase of $\left[\mathrm{N}\left(\mathrm{CH}_{3}\right)_{4}\right]_{2} \mathrm{ZnCl}_{4}$ 

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#### Abstract

The crystal structure of the threefold monoclinic lowtemperature phase of bis(tetramethylammonium) tetrachlorozincate(II), $\left[\mathrm{N}\left(\mathrm{CH}_{3}\right)_{4}\right]_{2} \mathrm{ZnCl}_{4}$, has been studied at 250 and 200 K . The $R$ values are 0.062 (6893 reflections, 3616 observed) and 0.064 (1907 reflections, 1558 observed) at 250 and 200 K respectively. The distortion with respect to the hightemperature phase is analyzed in terms of symmetry modes, and compared with the distortion in the incommensurate phase. The mode with $\sum_{3}$ symmetry responsible for the normal-incommensurate transition is still predominant in the distortion. Crystal data: $M_{r}=355 \cdot 5, \lambda($ Mo $K \alpha)=0.7107 \AA, P 2_{1} / n, Z=$ 12, $F(000)=2208$. At $250 \mathrm{~K}: a=8.953$ (3),$b=$ $15 \cdot 456$ (10), $c=36 \cdot 595$ (8) $\AA, \gamma=90 \cdot 19$ (4) ${ }^{\circ}, \quad V=$ 5064 (6) $\AA^{3}, \quad D_{x}=1.40 \mathrm{~g} \mathrm{~cm}^{-3}, \quad \mu=20.7 \mathrm{~cm}^{-1}$. At $200 \mathrm{~K}: a=8 \cdot 960$ (2), $b=15 \cdot 309$ (5), $c=36 \cdot 628$ (8) $\AA$, $\gamma=90.46(2)^{\circ}, V=5024(4) \AA^{3}, D_{x}=1.41 \mathrm{~g} \mathrm{~cm}^{-3}, \mu$ $=21 \cdot 0 \mathrm{~cm}^{-1}$.


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## 1. Introduction

Bis(tetramethylammonium) tetrachlorozincate (TMATC-Zn) is one of the most studied compounds of the $A_{2} B X_{4}$ family having a one-dimensional incommensurate (INC) phase. This compound undergoes five solid-solid phase transitions:

where the temperatures are those obtained from calorimetric measurements by Ruiz-Larrea, LopezEcharri \& Tello (1981).
Diffraction patterns of phases II, III and IV show superstructure reflections along the $\mathbf{c}^{*}$ direction that correspond to periodicities of $-5 c_{0}$ (INC), $5 c_{0}$ and $3 c_{0}$ respectively, $c_{0}$ being the lattice constant of phase I (Tanisaki \& Mashiyama, 1980). These satellite reflections are all weak and the structures can be considered small distortions of the parent phase I.
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The structure of phase I (basic structure, space group Pmen) was determined by Wiesner, Srivastava, Kennard, Di Vaira \& Lingafelter (1967). The structures of phase II [INC phase, superspace group $P(P m c n):(s 1 \overline{1})]$ and the ferroelectric phase III (lockin phase, space group $P 2_{1} c n$ ) have been reported by Madariaga, Zúñiga, Pérez-Mato \& Tello (1987).

Using phenomenological approaches, Mashiyama (1980) obtained the order-parameter symmetry characterizing the phase-transition sequence. The order parameter is associated with a modulation having the transformation properties of the irreducible representation $\sum_{3}$ with a wavevector $\mathbf{k}=$ $\alpha \mathbf{c}^{*}$, where $\alpha$ takes different values in each phase. In phase III the modulation wavevector locks at the value of $2 / 5 \mathbf{c}^{*}$, whereas in phase IV $\alpha$ takes the value 1/3.

The structural analysis of the INC and lock-in structures (phases II and III), reported in Madariaga, Zúñiga, Pérez-Mato \& Tello (1987), demonstrated that the main part of the distortion describing phases II and III has $\Sigma_{3}$ symmetry and is practically the same in both structures except for the change of wavevector of the modulation and a small increase of the global amplitude.

Phase IV is a threefold structure with space group $P 112_{1} / n$ (Tanisaki \& Mashiyama, 1980). In this paper, two structure determinations of this phase at 250 and 200 K respectively are presented. The double determination is motivated by the high disorder found at 250 K which decreases at 200 K . The structural distortion describing this phase with respect to phase I is decomposed in its contributing modes. The most relevant components are compared with those of the INC phase. Despite the change of symmetry, the predominance of a $\Sigma_{3}$ modulation which in broad terms still keeps the same internal structure as in the INC phase is also observed in this phase.

## 2. Experimental

Single crystals of TMATC-Zn were grown by slow evaporation at room temperature from an aqueous solution containing $\mathrm{N}\left(\mathrm{CH}_{3}\right)_{4} \mathrm{Cl}$ and $\mathrm{ZnCl}_{2}$ in stoichiometric proportions. The structure has been analyzed at 250 and 200 K , with diffraction intensities collected from two different crystals. The measurements at 250 K were performed with a CAD-4 diffractometer equipped with an $\mathrm{N}_{2}$-gas-flow cooling system (Cosier \& Glazer, 1986), whereas for those at 200 K , a Syntex $P 2_{1}$ diffractometer with an Enraf-Nonius low-temperature system was used. In both cases, when cooling through the III-IV transition, untwinned crystals were obtained. The stability of the measurements was checked periodically by measuring three standard reflections, and they were used to correct a mean intensity decay of 10 and $8 \%$

## Table 1. Summary of experimental data

|  | 250 K | 200 K |
| :---: | :---: | :---: |
| Crystal shape | Prismatic | Sphere |
| Crystal size (mm) | $0.1 \times 0.2 \times 0.15$ | 0.22 |
| Temperature stability (K) | 0.2 | 1 |
| Monochromator/filter | Graphite | Nb |
| Reflections for lattice constants | 25 ( $10<\theta<16$ ) | $15\left(10<\theta<18^{\prime}\right)$ |
| Scan | $\theta-2 \theta$ | $\theta 2 \theta$ |
| Scan speed ( min ${ }^{\prime}$ ) | 4.12-0.5 | 10.0-2.0 |
| Limits $h, k, l$ | $\pm 12.21 .39$ | $\pm 6,11,26$ |
| $(\sin \theta / \lambda)_{\text {max }}\left(\AA^{\prime}\right)$ | 0.70 | 0.367 |
| Absortion correction ( $A^{*}$ ) | None | 1.35 1.34 |
| Independent reflections | 6893 | 1907 |
| Observed reflections ( $I>3 \sigma$ ) | 3616 | 1558 |
| Weights | $1 / \sigma^{2}$ | $1 / \sigma^{2}$ |
| Maximum (shift/e.s.d.) | 0.86 | 0.23 |

(at 250 and 200 K respectively) at the end of the measurements, which is attributeed to radiation damage. Table 1 summarizes the relevant crystal data and data-collection information for both measurements.

## 3. Structure of phase IV

For the reduction of the X-ray measurments and the refinement of the structures, the XRAY system of programs (Stewart, Kruger, Ammon, Dickinson \& Hall, 1972) was used. The intensities were corrected for Lorentz-polarization and absorption factors. In the structure-factor calculations, scattering factors for $\mathrm{Zn}^{2+}, \mathrm{Cl}^{-}, \mathrm{N}$ and C atoms were taken from Cromer \& Mann (1968) and the anomalousscattering corrections for Zn and Cl atoms were those of Cromer \& Liberman (1970). H atoms were not considered in the refinements.

The structure at 250 K was refined by a blockdiagonal least-squares method with anisotropic thermal parameters for all atoms. We used as starting parameters for the Zn and Cl atoms those of phase I extended over three consecutive cells. During the initial stages of the refinement a soft restriction for the $\mathrm{Zn}-\mathrm{Cl}$ distances (prescribed to a value of $2 \cdot 23 \AA$ ) was used. Successive Fourier maps revealed all the N and C atoms. The final residuals based on $F$ were $R=0.062$ and $w R=0.063$ with $w=1 / \sigma^{2}(F)$. The goodness of fit was $2 \cdot 28$.* A final difference Fourier synthesis showed some peaks of $0.90 \mathrm{e}^{\AA^{-3}}$ near the $\mathrm{C}(123)$ and $\mathrm{C}(124)$ atoms, but attempts to refine the structure with a splitting for these atoms failed. The atomic parameters of phase IV of TMATC-Zn at 250 K are reported in Table 2.
The structure at 200 K was refined including anisotropic thermal parameters for the Zn and Cl

[^0]Table 2. Atomic parameters of phase IV at 250 K
The first appended numeral denotes the subcell of phase 1. E.s.d.'s are given in parentheses. $U_{\mathrm{cq}}=1 / 3$ (trace of the orthogonalized $U_{1 y}$, matrix).

|  | $x$ | $y$ | $y$ | $U_{\text {cq }}\left(\AA^{2}\right)$ |
| :---: | :---: | :---: | :---: | :---: |
| Zn (1) | $0 \cdot 2552$ (1) | $0 \cdot 40705$ (8) | 0.08153 (3) | 0.0402 (5) |
| $\mathrm{Cl}(11)$ | 0.2779 (3) | 0.4047 (2) | 0.02070 (9) | 0.064 (1) |
| $\mathrm{Cl}(12)$ | 0.2878 (3) | 0.5423 (2) | 0.1038 (1) | 0.077 (1) |
| $\mathrm{Cl}(13)$ | 0.4350 (3) | $0 \cdot 3228$ (2) | 0.1063 (1) | 0.071 (1) |
| $\mathrm{Cl}(14)$ | 0.0276 (3) | 0.3551 (2) | 0.09774 (9) | 0.069 (1) |
| N(11) | 0.2765 (7) | 0.0935 (5) | 0.0500 (2) | 0.040 (3) |
| C(111) | 0.249 (1) | 0.0932 (9) | 0.0911 (3) | 0.090 (7) |
| C(112) | $0 \cdot 240$ (1) | 0.0083 (8) | 0.0340 (3) | 0.072 (5) |
| C(113) | 0.180 (1) | 0.1626 (9) | 0.0312 (4) | 0.099 (7) |
| C(114) | 0.443 (1) | 0.1131 (9) | 0.0438 (4) | 0.098 (7) |
| N (12) | 0.2669 (8) | 0.8263 (5) | 0.1665 (2) | 0.053 (4) |
| C(121) | 0.251 (1) | 0.7320 (9) | 0.1540 (4) | 0.123 (8) |
| C(122) | 0.121 (2) | 0.8611 (9) | 0.1727 (6) | 0.17 (1) |
| C(123) | 0.345 (2) | 0.876 (1) | 0.1397 (6) | 0.25 (2) |
| C(124) | 0.346 (2) | 0.830 (1) | $0 \cdot 1990$ (6) | 0.27 (2) |
| $\mathrm{Zn}(2)$ | 0.2841 (1) | 0.40727 (7) | 0.41561 (3) | 0.0387 (5) |
| $\mathrm{Cl}(21)$ | 0.2658 (3) | 0.4089 (2) | 0.35430 (9) | 0.076 (1) |
| $\mathrm{Cl}(22)$ | 0.3239 (3) | 0.5424 (2) | 0.4378 (1) | 0.075 (1) |
| $\mathrm{Cl}(23)$ | 0.4781 (3) | $0 \cdot 3219$ (2) | 0.4323 (1) | 0.068 (1) |
| $\mathrm{Cl}(24)$ | 0.0692 (3) | 0.3546 (2) | 0.43914 (9) | 0.071 (1) |
| N (21) | 0.2532 (7) | 0.0955 (5) | 0.3835 (2) | 0.056 (4) |
| C(211) | $0 \cdot 246$ (2) | 0.0989 (9) | 0.4246 (4) | 0.116 (9) |
| C(212) | 0.204 (1) | 0.0074 (7) | 0.3705 (3) | 0.087 (6) |
| C(213) | 0.156 (2) | $0 \cdot 1648$ (9) | 0.3675 (4) | 0.123 (8) |
| C(214) | 0.416 (1) | 0.110 (1) | 0.3717 (4) | 0.124 (8) |
| N (22) | 0.2601 (7) | 0.8277 (4) | 0.4965 (2) | 0.040 (3) |
| C(221) | 0.405 (1) | 0.8691 (7) | 0.5092 (4) | 0.079 (6) |
| C(222) | 0.297 (1) | 0.7382 (8) | 0.4803 (4) | 0.096 (6) |
| C(223) | $0 \cdot 155$ (1) | 0.8178 (8) | 0.5279 (4) | 0.110 (7) |
| C(224) | 0.191 (1) | 0.8810 (8) | 0.4679 (4) | 0.091 (6) |
| Zn (3) | 0.2202 (1) | 0.40342 (7) | 0.74880 (4) | 0.0385 (5) |
| $\mathrm{Cl}(31)$ | 0.2062 (3) | 0.4090 (2) | 0.68757 (8) | 0.065 (1) |
| $\mathrm{Cl}(32)$ | 0.1866 (3) | 0.5358 (2) | 0.77431 (9) | 0.068 (1) |
| $\mathrm{Cl}(33)$ | 0.4483 (3) | 0.3527 (2) | 0.7657 (1) | 0.070 (1) |
| $\mathrm{Cl}(34)$ | 0.0387 (3) | 0.3139 (2) | 0.7696 (1) | 0.071 (1) |
| N(31) | 0.2270 (7) | 0.0910 (5) | 0.7155 (2) | 0.044 (3) |
| C(311) | 0.269 (1) | 0.0034 (8) | 0.7017 (4) | 0.095 (7) |
| C(312) | 0.257 (1) | 0.0941 (8) | 0.7554 (3) | 0.092 (7) |
| C(313) | 0.064 (1) | 0.1093 (9) | 0.7072 (3) | 0.091 (6) |
| C(314) | 0.322 (1) | 0.1589 (8) | 0.6968 (4) | 0.091 (6) |
| N(32) | 0.2216 (7) | 0.8244 (5) | 0.8312 (2) | $0 \cdot 046$ (3) |
| C(321) | 0.068 (1) | 0.8478 (8) | 0.8414 (4) | $0 \cdot 116$ (8) |
| C(322) | 0.320 (2) | 0.8311 (9) | 0.8641 (5) | 0.151 (9) |
| C(323) | 0.279 (2) | 0.885 (1) | 0.8028 (4) | 0.133 (9) |
| C(324) | $0 \cdot 218$ (1) | 0.7338 (8) | 0.8162 (4) | 0.127 (9) |

atoms but isotropic ones for N and C atoms. The final residual values obtained are $R=0.064$ and $w R$ $=0.066$. ${ }^{*}$

The above-mentioned peaks in the difference Fourier synthesis at 250 K were at typical N-C distances of the $\mathrm{N}(12)$ atom. This fact, together with the large thermal parameters found for $\mathrm{C}(121)$, $\mathrm{C}(122), \mathrm{C}(123)$ and $\mathrm{C}(124)$, points to a disorder for this tetramethylammonium group. The residual peaks persist at 200 K and even if the thermal parameters decrease, they continue to have anomalously large values of 0.16 and $0.15 \AA^{2}$ for $\mathrm{C}(123)$ and $\mathrm{C}(124)$. Relatively large values for the $U_{11}$ thermal parameter are also found in the other organic groups indicating the existence of a disorder which in general is greater for the ions associated with $\mathrm{N}(12)$, $\mathrm{N}(22)$ and $\mathrm{N}(32)$. This result can be understood if we consider that in the structure the Cl atoms form two types of cages, one larger than the other, in which

[^1]the tetramethylammonium ions are located. A similar result is found in the threefold structure of $\mathrm{Rb}_{2} \mathrm{ZnBr}_{4}$ where the Br cavity of the $\mathrm{Rb}(2)$ is considerably larger than that of $\mathrm{Rb}(1)$ (Hogervorst \& Helmholdt, 1988).

## 4. Discussion

Projections of the structure (with C atoms excluded) in the $a c$ plane at 250 K and at room temperature are shown in Fig. 1. Bond lengths and angles for phase IV at 250 K are given in Table 3. The most interesting feature of the structure is its distortion when referred to the 'basic' structure (phase I). If the small difference of the monoclinic angle from $90^{\circ}$ is neglected, the structural distortion can be calculated by subtracting the coordinates of phase I given in Table 3 of Madariaga, Zúñiga, Pérez-Mato \& Tello (1987) (in three consecutive cells) from those in Table 2. As a general result it can be concluded that for all atoms ihe $x$ component of the atomic displacements is the most important. In the case of the atoms lying on the mirror plane in phase $I$ it is even an order of magnitude higher than the $y$ and $z$ components.
A better insight into the structure can be obtained if the structural distortion is decomposed in terms of symmetry modes of the basic structure (phase I) as reported previously in other cases (Pérez-Mato, Gaztelua, Madariaga \& Tello, 1986). This allows a more quantitative comparison with phases II(INC)

(a)

(b)

Fig. 1. (a) ac projection of the basic structure and (b) the threefold-modulated monoclinic phase IV of TMATC-Zn. The C atoms of the tetramethylammonium groups are omitted for clarity.

Table 3. Distances $(\AA)$ and angles $\left({ }^{\circ}\right)$ of phase IV at 250 K

| E.s.d.s are given in parentheses. |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
| $\mathrm{Zn}(1)-\mathrm{Cl}(11) \quad 2.236$ (4) | $\mathrm{Zn}(2)-\mathrm{Cl}(21)$ | 2.250 (4) | $\mathrm{Zn}(3)-\mathrm{C}(31)$ | $2 \cdot 246$ (3) |
| $-\mathrm{Cl}(12) \quad 2.262(4)$ | - $\mathrm{Cl}(22)$ | 2.268 (4) | - $\mathrm{Cl}(32)$ | $2 \cdot 270$ (3) |
| - $\mathrm{Cl}(13) \quad 2.263(3)$ | $-\mathrm{Cl}(23)$ | $2 \cdot 269$ (3) | - $\mathrm{Cl}(33)$ | $2 \cdot 276$ (3) |
| - $\mathrm{Cl}(14) \quad 2.266$ (3) | - $\mathrm{Cl}(24)$ | $2 \cdot 258$ (3) | $-\mathrm{Cl}(34)$ | $2 \cdot 263$ (3) |
| $\mathrm{N}(11)-\mathrm{C}(111) \quad 1.52$ (1) | $\mathrm{N}(21)-\mathrm{C}(211)$ | 1.50 (2) | $\mathrm{N}(31)-\mathrm{C}(311)$ | 1.49 (1) |
| - $\mathrm{C}(112) \quad 1.48$ (1) | -C(212) | 1.51 (1) | - $\mathrm{C}(312)$ | 1.49 (1) |
| - C(113) 1.54 (2) | -C(213) | 1.50 (2) | - $\mathrm{C}(313)$ | 1.52 (1) |
| -C(114) 1.53 (1) | -C(214) | 1.53 (1) | - C (314) | 1.51 (1) |
| $\mathrm{N}(12)-\mathrm{C}(121) \quad 1.53$ (2) | $\mathrm{N}(22)-\mathrm{C}(221)$ | 1.52 (I) | $\mathrm{N}(32)-\mathrm{C}(321)$ | 1.47 (1) |
| - C(122) 1.43 (2) | -C(222) | 1.54 (1) | - $(322)$ | 1.49 (2) |
| $-\mathrm{C}(123) \quad 1.43$ (2) | -C(223) | 1.50 (2) | -C(323) | 1.49 (2) |
| $-\mathrm{C}(124) \quad 1.38$ (2) | -C(224) | 1.47 (1) | - C (324) | $1 \cdot 50$ (2) |
| $\mathrm{Cl}(11)-\mathrm{Zn}(1)-\mathrm{Cl}(12)$ | 111.2(1) | C(211) | - $\mathrm{N}(21)-\mathrm{C}(212)$ | 109.6 (9) |
| $\mathrm{Cl}(11)-\mathrm{Zn}(1)-\mathrm{Cl}(13)$ | 109.9 (1) | C(211) | - $\mathrm{N}(21)-\mathrm{C}(213)$ | 109.9 (9) |
| $\mathrm{Cl}(11)-\mathrm{Zn}(1)-\mathrm{Cl}(14)$ | 109.7 (1) | C(211) | N(21)-C(214) | 108.4 (9) |
| $\mathrm{Cl}(12)-\mathrm{Zn}(1)-\mathrm{Cl}(13)$ | 107.3 (1) | C(212) | $\mathrm{N}(21)-\mathrm{C}(213)$ | 110.5 (8) |
| $\mathrm{Cl}(12)-\mathrm{Zn}(1)-\mathrm{Cl}(14)$ | $110 \cdot 3$ (1) | C(212) | $\mathrm{N}(21)-\mathrm{C}(214)$ | 108.5 (8) |
| $\mathrm{Cl}(13)-\mathrm{Zn}(1)-\mathrm{Cl}(14)$ | $109 \cdot 4$ (1) | $\mathrm{C}(213)$ | - $\mathrm{N}(21)-\mathrm{C}(214)$ | 110.0 (9) |
| $\mathrm{Cl}(21)-\mathrm{Zn}(2)-\mathrm{Cl}(22)$ | 110.0 (1) | C(221) | - $\mathrm{N}(22)-\mathrm{C}(222)$ | 108.1 (7) |
| $\mathrm{Cl}(21)-\mathrm{Zn}(2)-\mathrm{Cl}(23)$ | $109 \cdot 3$ (1) | $\mathrm{C}(221)$ | - $\mathrm{N}(22)-\mathrm{C}(223)$ | 110.2 (8) |
| $\mathrm{Cl}(21)-\mathrm{Zn}(2)-\mathrm{Cl}(24)$ | 108.8 (1) | $\mathrm{C}(221)$ | - $\mathrm{N}(22)-\mathrm{C}(224)$ | 110.1 (8) |
| $\mathrm{Cl}(22)-\mathrm{Zn}(2)-\mathrm{Cl}(23)$ | 108.7 (1) | C(222) | - $\mathrm{N}(22)-\mathrm{C}(223)$ | 110.0 (8) |
| $\mathrm{Cl}(22)-\mathrm{Zn}(2)-\mathrm{Cl}(24)$ | 109.1 (1) | $\mathrm{C}(222)$ | $\mathrm{N}(22)-\mathrm{C}(224)$ | 108.7 (8) |
| $\mathrm{Cl}(23)-\mathrm{Zn}(2)-\mathrm{Cl}(24)$ | 109.9 (1) | $\mathrm{C}(223)$ | $\mathrm{N}(22)-\mathrm{C}(224)$ | 109.8 (8) |
| $\mathrm{Cl}(31)-\mathrm{Zn}(3)-\mathrm{Cl}(32)$ | 111.6 (1) | C(311) | N(31)-C(312) | 108.5 (8) |
| $\mathrm{Cl}(31)-\mathrm{Zn}(3)-\mathrm{Cl}(33)$ | 109.6 (1) | C(311) | N(31)-C(313) | 110.2 (8) |
| $\mathrm{Cl}(31)-\mathrm{Zn}(3)-\mathrm{Cl}(34)$ | 108.6 (1) | C(311) | N(31)-C(314) | 109.6 (8) |
| $\mathrm{Cl}(32)-\mathrm{Zn}(3)-\mathrm{Cl}(33)$ | $108 \cdot 6$ (1) | C(312) | N(31)-C(313) | $111 \cdot 3$ (8) |
| $\mathrm{Cl}(32)-\mathrm{Zn}(3)-\mathrm{Cl}(34)$ | 108.4 (1) | C(312) | N(31)-C(314) | 108.8 (8) |
| $\mathrm{Cl}(33)-\mathrm{Zn}(3)-\mathrm{Cl}(34)$ | 110.0 (1) | C(313) | $\mathrm{N}(31)-\mathrm{C}(314)$ | 108.5 (8) |
| $\mathrm{C}(111)-\mathrm{N}(11)-\mathrm{C}(112)$ | 110.7 (8) | $\mathrm{C}(321)$ | N(32)-C(322) | 109 (1) |
| $\mathrm{C}(111)-\mathrm{N}(11)-\mathrm{C}(113)$ | $110 \cdot 6$ (8) | C(321) | - $\mathrm{N}(32)-\mathrm{C}(323)$ | 110.0 (9) |
| $\mathrm{C}(111)-\mathrm{N}(11)-\mathrm{C}(114)$ | 107.6 (8) | $\mathrm{C}(321)$ | - $\mathrm{N}(32)-\mathrm{C}(324)$ | 107.8 (8) |
| $\mathrm{C}(112)-\mathrm{N}(11)-\mathrm{C}(113)$ | 108.7 (8) | $\mathrm{C}(322)$ | - $\mathrm{N}(32)-\mathrm{C}(323)$ | 108.4 (9) |
| $\mathrm{C}(112)-\mathrm{N}(11)-\mathrm{C}(114)$ | 109.1 (8) | $\mathrm{C}(322)$ | N(32)-C(324) | 111.6 (9) |
| $\mathrm{C}(113)-\mathrm{N}(11)-\mathrm{C}(114)$ | 110.1 (8) | $\mathrm{C}(323)$ | N(32)-C(324) | 110 (1) |
| $\mathrm{C}(121)-\mathrm{N}(12)-\mathrm{C}(122)$ | 108.9 (9) |  |  |  |
| $\mathrm{C}(121)-\mathrm{N}(12)-\mathrm{C}(123)$ | 111 (1) |  |  |  |
| $\mathrm{C}(121)-\mathrm{N}(12)-\mathrm{C}(124)$ | 110 (1) |  |  |  |
| $\mathrm{C}(122)-\mathrm{N}(12)-\mathrm{C}(123)$ | 111 (1) |  |  |  |
| $\mathrm{C}(122)-\mathrm{N}(12)-\mathrm{C}(124)$ | 108 (1) |  |  |  |
| $\mathrm{C}(123)-\mathrm{N}(12)-\mathrm{C}(124)$ | 109 (1) |  |  |  |

and III(lock-in). In the present case, the analysis for the atoms lying on the mirror plane in phase I can be reduced to express atom displacements in the form:

$$
\begin{equation*}
U_{\alpha}(L)=U_{0 \alpha}+U_{1 \alpha} \cos \left[(2 \pi / 3)(z+L)+\varphi_{1 \alpha}\right] . \tag{1}
\end{equation*}
$$

This expression describes the atomic displacements of the asymmetric unit of phase IV. Equation (1) is considered for each set of three atoms which are translationally equivalent in phase I. In (1), $L=0,1$, 2 stands for the three successive subcells, $\alpha=x, y, z$ refers to the three components and $z$ represents the $z$ component of the unique atom in the asymmetric unit of phase I to which the three atoms considered reduce when passing to phase I. Thus, the three atomic displacements are explained as the sum of a homogeneous distortion plus a sinusoidal modulation with wavevector $q=(1 / 3) \mathbf{c}^{*}$. The homogeneous distortion $\mathbf{U}_{0}$, when considered for all the atoms in the subcell, has symmetry $A_{1 g}$ in the case of the $y$ and $z$ components, while the $x$ component corresponds to a $B_{3 g}$ mode. The sinusoidal modulation along the $a$ axis, given by the amplitudes $U_{1 x}$ and the phases $\varphi_{1 x}$, describes a $\Sigma_{3}$ mode, whereas
those corresponding to the $y$ and $z$ directions have the transformation properties of the irreducible representation $\Sigma_{4}$. This explains the large values of the atomic displacements along the $a$ axis. The distortion mode of symmetry $\Sigma_{3}$ is predominant in phase IV since $\Sigma_{3}$ is in fact the symmetry of the order parameter which is responsible for the whole phasetransition sequence in this compound (Mashiyama, 1980).

In the case of the atoms out of the mirror plane in phase I, the decomposition (1) is incomplete for a description in terms of symmetry modes. It must be substituted by:

$$
\begin{align*}
U_{\alpha}(L)= & U_{0 \alpha}+U_{1 \alpha} \cos \left[(2 \pi / 3)(z+L)+\varphi_{1 \alpha}\right] \\
& +U_{2 \alpha} \cos \left[(4 \pi / 3)(z+L)+\varphi_{2 \alpha}\right]+U_{3 \alpha} . \tag{2}
\end{align*}
$$

The displacement $\mathbf{U}_{0}$ is an homogeneous distortion of $A_{1 g}$ symmetry. The other homogeneous term $\mathbf{U}_{3}$ represents a $B_{3 g}$ distortion, whereas the sinusoidal terms $\mathbf{U}_{1}$ and $\mathbf{U}_{2}$ correspond to distortions of symmetries $\Sigma_{3}$ and $\Sigma_{4}$ respectively. None of these modes have their corresponding atomic displacements restricted in the number of non-zero components. A complete determination of their amplitudes and phases also requires consideration of the displacements $\mathbf{U}^{\prime}(1)$ of the three atoms in phase I symmetry related by the mirror plane $m_{x}$ with those considered in expression (2). These displacements should satisfy for the $x$ component:

$$
\begin{align*}
U_{x}^{\prime}(L)= & -U_{0 x}+U_{1 x} \cos \left[(2 \pi / 3)(z+L)+\varphi_{1 x}\right] \\
& -U_{2 x} \cos \left[(4 \pi / 3)(z+L) \varphi_{2 x}\right]+U_{3 x} \tag{3}
\end{align*}
$$

while for the $y$ and $z$ components:

$$
\begin{align*}
U_{y, z}^{\prime}(L)= & \left.U_{0 y, z}-U_{1 y, z} \cos \left[(2 \pi / 3)(z+L)+\varphi_{1 y, z}\right)\right] \\
& +U_{2 y, z} \cos \left[(4 \pi / 3)(z+L)+\varphi_{2 y, z}\right]-U_{3 y, z} . \tag{4}
\end{align*}
$$

Expressions (3) and (4) can be derived from the symmetry properties, which by definition are associated to each term in (2).
The final results along the $a$ axis for the ions in the inorganic tetrahedra and the N atoms are given in Table 4. The $\Sigma_{3}$ modulation clearly constitutes the main part of the whole structural distortion. The same happens along the other two axes, where only atom $\mathrm{Cl}(3)$ has a $\Sigma_{3}$ contribution. A similar result has been observed in phase III. The $\Sigma_{3}$ modulation in the INC phase (II) is also given in Table 4 for comparison. It can be observed that the structure (the polarization vector) of the $\Sigma_{3}$ distortion in phase IV is not only essentially the same at the two temperatures investigated, but it also coincides with that of the INC phase. In other words, the phases $\varphi_{1}$ are maintained practically invariant, as was observed for phase III (Madariaga, Zúñiga, Pérez-Mato \& Tello, 1987). Broadly speaking, the difference between the three modulations is reduced to a single common

Table 4. Amplitudes $\left(\times 10^{4}\right)$ and phases $(\times 2 \pi)$ of the harmonics describing the distortion along the a axis at 250 K (first line) and 200 K (second line)

Values for the $\mathrm{Cl}(4)$ atom were obtained from those of $\mathrm{Cl}(3)$ after application of the $\sigma_{x}$ symmetry operation. The $\Sigma_{3}(x)$ mode of the INC phase is also included for comparison.

| Zn | $A_{18}$ | $\sum_{\left(U_{1} / \varphi_{1}\right)}^{\sum_{3}}$ | $\sum_{\left(U_{2} / \varphi_{2}\right)}$ | $B_{3 g}$ | $\begin{gathered} \sum_{3}(\text { INC }) \\ \left(U_{1} / \varphi_{1}\right) \end{gathered}$ |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | 390/0.681 |  | 21 | 92/0.658 |
|  |  | 474/0.681 |  | 44 |  |
| $\mathrm{Cl}(1)$ |  | 443/0.836 |  | -1 | 109/0.817 |
|  |  | 563/0.842 |  | -13 |  |
| $\mathrm{Cl}(2)$ |  | 820/0.689 |  | 161 | 233/0.692 |
|  |  | 968/0.683 |  | 187 |  |
| $\mathrm{Cl}(3)$ | -15 | 252/0.519 | 7/0.347 | -5 | 57/0.464 |
|  | -24 | 334/0.542 | 6/0.347 | -1 |  |
| $\mathrm{N}(1)$ |  | 286/0.911 |  | 22 | 178/0.903 |
|  |  | 376/0.903 |  | 40 |  |
| $\mathrm{N}(2)$ |  | 282/0.856 |  | -5 | 232/0.854 |
|  |  | 370/0.853 |  | -3 |  |

factor which represents the amplitude of the global distortion. These results demonstrate that the predominant structural modulation corresponding to the primary-order parameter is essentially the same in phases II, III and IV. The internal structure of the modulation is maintained despite the strong symmetry changes, except for the change of wavevector and the global amplitude, which increases monotonically as the temperature decreases. The symmetry changes are due to the variation of modulation wavevector and the change of the initial phase of the modulation. This confirms from a microscopic viewpoint the assumptions made in a Landau model of these phases, where the different phases are essentially described by the amplitude and phase of a single complex parameter (order parameter) and the modulation wavevector. The relevance of secondary parameters is shown to be rather small.

Some of the data in Table 4 are implicitly correlated because of the rigidity of the inorganic $\mathrm{ZnCl}_{4}$ tetrahedra. It is therefore interesting to describe the atomic displacements corresponding to these atoms in terms of small rotations and translations of these tetrahedra as rigid bodies. A fit along these lines by means of a least-squares program was successful with residues which typically amounted to $10 \%$. The rotations obtained for the tetrahedra in three consecutive subcells can be fitted to an expression similar to (1):

$$
\begin{equation*}
R_{\alpha}(L)=R_{0 \alpha}+R_{1 \alpha} \cos \left[(2 \pi / 3) L+\varphi_{1 \alpha}^{R}\right] . \tag{5}
\end{equation*}
$$

Something analogous can be applied to the tetrahedra translations. The results are compared in Table 5 with those obtained for the INC phase at 286 K . The homogeneous terms $R_{0 . x}, T_{0 y}, T_{0 z}$ have $A_{1 g}$ symmetry, whereas $R_{0 y}, R_{0 z}, \tilde{i}_{0 x}$ have $B_{3 g}$ symmetry. The modulation terms $R_{1 y}, R_{1 z}, T_{1 x}$ correspond to $\Sigma_{3}$

Table 5. Amplitudes and phases $(\times 2 \pi)$ of the harmonic rotations and translations describing the distortion of rigid $\mathrm{ZnCl}_{4}$ ions at 250 K (first line) and 200K (second line)

The values for the INC phase are also included for comparison.

|  | $R_{0}$ | $R_{1} / \varphi_{1}$ | INC phase $R_{1} / \varphi_{1}$ |
| :---: | :---: | :---: | :---: |
| $R_{\text {r }}()$ | $0 \cdot 69$ | 1.63/0.622 | $0 \cdot 01^{\prime} 0903$ |
|  | 1.35 | 3.30/0.623 |  |
| $R_{v}\left({ }^{\prime}\right)$ | 0.64 | 5.53/0.523 | $1.97 \% 0.512$ |
|  | 1.22 | 7.03/0.526 |  |
| $R_{\text {: }}($ ) | -2.39 | 11.75/0.338 | $7 \cdot 44 / 0 \cdot 360$ |
|  | -2.70 | 12.92/0.340 |  |
|  | $T_{0}$ | $T_{1} / \varphi_{1}$ | $T_{1}{ }^{\prime} \varphi_{1}$ |
| $T_{\lambda}\left(\times 10^{4}\right)$ | 36 | 373/0.763 | 180/0.774 |
|  | 43 | 476/0.765 |  |
| $T_{1}\left(\times 10^{4}\right)$ | -14 | 29/0.187 | $1{ }^{\prime} 0.660$ |
|  | -19 | 50/0.185 |  |
| $T_{:}\left(\times 10^{4}\right)$ | 1 | 10/0.510 | $3 / 0.372$ |
|  | 0 | 13/0.505 |  |

symmetry and $R_{1 x}, T_{1 y}, T_{1 z}$ to $\sum_{4}$. As before, the predominance of those terms corresponding to the primary-order parameter is clear. The phases of the sinusoidal terms typically vary less than a degree from 250 to 200 K , and in the case of the $\Sigma_{3}$ terms they essentially coincide with those observed in the INC phase. The phases corresponding to the $\Sigma_{4}$ terms have, however, no clear correlation with those observed in the INC phase. The amplitudes of these terms in the INC phase are practically two orders of magnitude smaller than those of the $\Sigma_{3}$ distortion. Hence the present result seems to indicate that these values have little physical significance and were probably an artefact of the refinement process.

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[^0]:    * Lists of structure factors and anisotropic thermal parameters at 200 and 250 K , and bond lengths, bond angles and atomic parameters at 200 K have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 51876 ( 37 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CHI 2HU, England.

[^1]:    * See deposition footnote

